Conference Abstract

Adsorption of Cadmium ion from aqueous solutions by stem and leaf of *Parthenium hysterophorus*

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Abstract

Cadmium is a toxic heavy metal causing risks for living organisms even at low levels of its presence. Poor treatment of Cd^{2+} from contaminated sources leads to high risks for human health causing health disorders like rheumatoid arthritis, kidney failure and allergies [1]. *Parthenium hysterophorus* is a non-living biomass of low cost, environment friendly bio sorbent and can be considered as a very effective adsorbent for the removal of heavy metals like Cd. Studies have been carried out on activated carbon prepared from parthenium for elimination of Cd and Ni from contaminated wastewater [2]. In this report we use dried plant parts of *Parthenium hysterophorus* as an adsorbent for the removal of Cd^{2+} ions from aqueous solution. The extent of adsorption of Cd^{2+} ions by stem and leaf parts of the *Parthenium hysterophorus* is analyzed by using acid and base activation, weight variation and concentration variation. Anodic Stripping Voltammetry (ASV) via standard addition technique was used in the quantitative determination of Cd^{2+} samples.

Results show that the percentage adsorption of the Cd^{2+} ranges from 41.31 to 99.59. Different operational parameters observed during the process of investigation reveal that the adsorbent dosage, initial adsorbate concentration, activated adsorbent dosage govern the overall process of adsorption. The effective concentration of Cd^{2+} ion and the optimum dosage of *Parthenium hysterophorus* were calculated as 100 μ M and 50 mg of acid activated leaf powder respectively. This results suggest that *Parthenium hysterophorus* can be used effectively to remove Cd^{2+} ions from wastewater.

Keywords: Heavy metal toxicity, Wastewater treatment, Parthenium hysterophorus

References

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